

Using UHPLC/MS (TOF) for Detection of Drugs and Metabolites in Urine following Optimized Enzymatic Hydrolysis Conditions

Background

 β -D-Glucuronide glucuronosohydrolase enzymes, commonly shortened to β -glucuronidase, play an important role in the analysis of biological fluids for the presence of metabolites for drug screening and drug metabolism studies. β -Glucuronidase hydrolyzes glucuronide metabolites back to the native parent drug by the general pathway shown in **Figure 1**.¹ Hydrolysis using β -glucuronidase is often necessary when extensive drug metabolism complicates drug detection in biological samples. For liquid chromatographic analytical methods, the highly-polar glucuronide metabolites that form cannot be easily retained in reversed-phase chromatographic separation, thus hindering the quantitative ability of the analytical method. Hydrolysis of glucuronide metabolites is necessary for gas chromatographic methods. This also commonly includes the addition of secondary derivatization for effective analysis and detection.

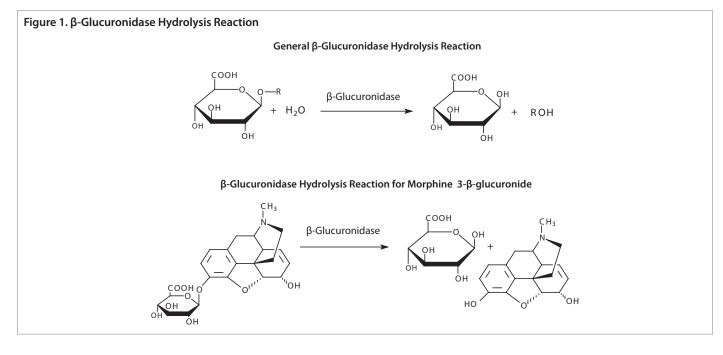
The enzyme is ubiquitous throughout living systems because of its important role in carbohydrate metabolism. Although there are numerous β -glucuronidase enzymes available, each enzyme has optimal conditions for hydrolysis of glucuronide metabolites. Variables such as β -glucuronidase enzyme concentration, digestion pH, incubation time, and temperature all play important roles for effective hydrolysis of glucuronide metabolites.

Goals of the Study

The goal of the study presented here was to determine optimal conditions for β -glucuronidase from various species toward hydrolysis of different compound classes with subsequent analysis by UHPLC/MS (TOF). It will address some of the critical variables that come into play when choosing an appropriate β -glucuronidase enzyme and incubation conditions to perform effective hydrolysis of glucuronide metabolites to the parent drug form. In this study, a set of compounds that are commonly hydrolyzed prior to analysis was used to evaluate the conditions for effective hydrolysis over a range of sources and forms of β -glucuronidase. Various strains of β -glucuronidase enzymes were evaluated. Five drug classes covering opioids, benzodiazepines, steroids, cannabinoids, and pain management drugs and eleven β -glucuronidases that differ in source, type, or formulation were included in the study (**Tables 1** and **2**).

Table 1. Compound Classes Included in the Study

Class	Representative Compounds
Opioids	Buprenorphine , Norbuprenorphine, Codeine, Hydromorphone, Oxymorphone, Morphine
Benzodiazepines	Lorazepam, Oxazepam, Temazepam
Steroid	Testosterone
Pain Management	Tapentadol
Cannabinioid	(-)-11-nor-9-Carboxy-Ƽ-THC



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Table 2. β-Glucuronidase, Sources, Forms, and Concentrations

G0251 Bovine liver, Type B-1 solid 1,644,000 U/g 30 mg/mL, pH 5 G0376 Bovine liver, Type B-3 solid 2,525,000 U/g 20 mg/mL, pH 5	
G0376 Boyine liver. Type B-3 solid 2,525,000 U/g 20 ma/ml , pH 5	
G0751Burgundy snail (Helix pomatia) Type H-1Iyophilized powder3,107,000 U/g16 mg/mL, pH 5	
G1512 Burgundy snail (<i>Helix pomatia</i>) Type H-5 lyophilized powder 978,500 U/g 51 mg/mL, pH 5	
G8295Escherichia coli, recombinantIyophilized powder11,458,000 U/g4.5 mg/mL, pH 6	
G0799Escherichia coli, recombinantaqueous glycerol solution10,000,000 U/g2.5 mg/mL, pH 6	
G7396 Escherichia coli, Type IX-A lyophilized powder 1,524,575 U/g 33 mg/mL, pH 6	
G7646Escherichia coli, Type VII-AIyophilized powder19,600,000 U/g2.5 mg/mL, pH 5	
G4259 Garden snail (<i>Helix aspersa</i>), Type HA-4 partially purified powder 602,600 U/g 83 mg/mL, pH 5	
G2174 Limpet (<i>Patella vulgata</i>) aqueous solution ≥85,000 U/mL aqueous solution w	vas used as is
G8132Limpet (Patella vulgata), Type L-IIIyophilized powder1,039,000 U/g48 mg/mL, pH 5	

Experimental

Preparation of Glucuronide Standards and Spiked Urine Samples

A stock solution of 10 μ g/mL of each glucuronide metabolite was prepared in water. Human urine was spiked at 200 ng/mL with each glucuronide metabolite.

Preparation of β-Glucuronidase Solutions

Lyophilized forms of β -glucuronidase were prepared in the following manner: Stock solutions of 50 kU/mL of each β -glucuronidase were prepared in a 100 mM ammonium acetate buffer at pH 5.0 and 6.0. Solutions were kept at 4 °C when not in use. Liquid forms were used as supplied.

Sample Digestion with β -Glucuronidase

Spiked stock urine samples were prepared at 10 mL volumes for multiple replicate sampling. Urine samples were spiked at 200 ng/mL with all glucuronide metabolites of test compounds. To the 10 mL urine samples, 2 mL of each corresponding pH buffer was added followed by 2 mL of β -glucuronidase solution. The resultant concentration of β -glucuronidase was 10 kU/mL per urine sample.

Urine samples were then digested at 60 °C in a convection oven for 120 minutes. Samples were removed from the oven and allowed to cool. A 1.4 mL aliquot was removed for further sample preparation using solid phase extraction (SPE) followed by analysis by UHPLC/MS (TOF).

Hydrolysis Variables

As part of this optimization study, experiments were conducted to evaluate the impact of incubation time, incubation temperature, pH, and enzyme concentration on the conversion of glucuronide metabolites to their corresponding parent drugs. Although these conditions are an important portion of the enzymatic conversion, full experimental details are beyond the scope of this report. The range of incubation times, temperatures, pH, and enzyme concentration tested are as follows:

- Incubation time: 30 minutes, 60 minutes, 120 minutes
- Incubation temperature: 45 °C, 55 °C, 60 °C, 65 °C
- Incubation pH: 4.5, 5, 6
- β-Glucuronidase concentration: 10 kU/mL and 50 kU/mL

Incubation Time

A rapid conversion of the glucuronide metabolite to the parent drug was observed for the short 30 minute incubation time for a majority of the enzymes tested. Conversion was optimal at the 120 minute incubation time for all drug/ β -glucuronidase combinations. Subsequent experiments were standardized on a 120 minute incubation time.

Incubation Temperature and Enzyme Concentration

Incubation temperature played an important role in enzymatic activity. The role of β -glucuronidase enzyme concentration was not as significant a rate limiting step as time or temperature. A significant increase in conversion was not observed between the 10 kU/mL and 50 kU/mL at the 120 minute incubation time study. The 10 kU/mL enzyme concentration was the lowest level incorporated. In the case of incubation temperatures above 65 °C, there was a noted decrease in glucuronide conversion with the 50 kU/mL enzyme concentration. Further study centered on 10 kU/mL enzyme concentration.

Incubation pH

Of all the digestion conditions evaluated, incubation pH had the most significant impact on effectiveness of β -glucuronidase activity. Because pH had such a major impact, the bulk of the results reported in this study are focused around the role of sample pH and the effectiveness of the various β -glucuronidase enzymes. The experimental conditions for pH studies were as follows:

- Incubation time: 120 minutes
- Incubation temperature: 60 °C
- β-Glucuronidase concentration: 10 kU/mL
- Incubation pH: 4.5, 5.0, 6.0
- Analyte concentration in urine: 200 ng/mL







Solid Phase Extraction (SPE) Method

SPE removes excess enzyme that can interfere with the analysis and foul the HPLC columns. Samples were processed by SPE using a mixed mode cation exchange/hydrophobic stationary phase. Native (parent) drugs were eluted from the SPE tube using a basic acetonitrile elution solvent. All sample processing was conducted in replicate (n=4).

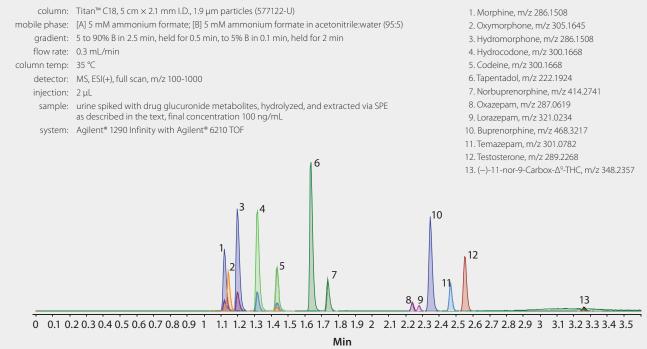
- SPE tube/cartridge: Supel[™]-Select SCX, 60 mg/3 mL (54241-U)
- Conditioning step: 1 mL 1% formic acid in acetonitrile followed by 1 mL water
- Sample loading: 1.4 mL spiked and hydrolyzed urine
- Washing step: 2 mL water followed by 1 mL 25% (v/v) methanol in water

- Elution: 1 mL 10% (w/v) ammonium hydroxide in acetonitrile
- Sample post-treatment: evaporate under nitrogen and reconstitute in 1 mL water

UHPLC/MS (TOF) Conditions

All analyses were conducted using gradient elution on a Titan[™] C18 1.9 µm UHPLC column. Glucuronide metabolite standards of the class compounds were spiked into control urine samples for determination of hydrolysis conversion. This known concentration of glucuronide metabolite in the urine sample was then used to calculate conversion to the parent drug. Standard solutions of native parent drug (10– 300 ng/mL in water) were used to develop calibration curves on the UHPLC MS/TOF system. Chromatographic conditions appear in **Figure 2**.

Figure 2. LC/MS (TOF) Analysis of Drugs and Their Glucuronide Metabolites on Titan[™] C18 after Solid Phase Extraction (SPE) using Supel-Select SCX following β-Glucuronidase Enzyme Digestion



Results

The optimized enzyme hydrolysis conditions for each drug are summarized in **Table 3**. A chromatogram of the test compounds hydrolyzed and extracted from urine, separated on the Titan[™] C18 column, and analyzed by LC/MS (TOF) is shown in **Figure 2**.

The critical pair of codeine and hydrocodone had sufficient resolution under these gradient conditions. This was extremely important since these compounds are isobaric. No sample matrix interference was observed which demonstrates the effectiveness of the sample preparation method.

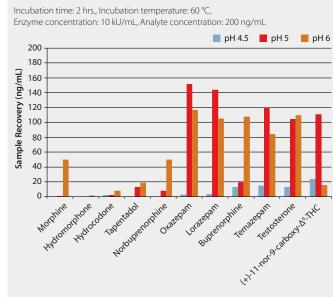
Effect of pH on Hydrolysis by β-Glucuronidase

The pH used during the digestion was the most significant factor of all design variables. For example, **Figure 3** shows glucuronide hydrolysis under the three pH conditions using the *E. coli* Type IX-A enzyme. Only marginal hydrolysis occurred at pH 4.5 for all compound classes, but pH 5 and pH 6 had significant hydrolysis. Hydrolysis above pH 5 was dependent on the compound class. The benzodiazepines, pain management, and steroid test compounds had substantial conversion above pH 5, while the opioids had only marginal conversion across any pH value. This would indicate the *E. coli* Type IX-A enzyme would not be suitable as a general hydrolysis for glucuronide metabolites but may be suitable for class-specific compounds.





Figure 3. *Escherichia coli* β-Glucuronidase Type IX-4 (G7396): Effect of pH on Hydrolysis Efficiency

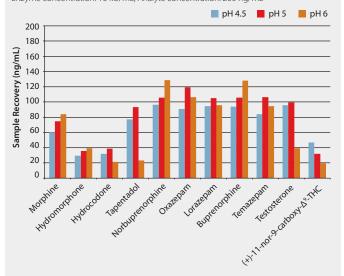


In comparison, effect of pH on limpet (*Patella vulgata*) β -glucuronidase Type L-II hydrolysis is shown in **Figure 4**. Hydrolysis was observed for all classes of compounds across all incubation pH values. Optimal hydrolysis for all classes was observed at pH 5, but this was not necessarily the optimal pH for each individual class. The benzodiazepines had optimal hydrolysis at pH 5 while the opioids had optimal hydrolysis at pH 6. These results would suggest the *Patella vulgata* Type L-II β -glucuronidase may be suitable as a general hydrolysis enzyme independent of compound class.

Table 3. Optimized Hydrolysis Conditions for Each Analyte

Figure 4. Limpet (*Patella vulgata*) β-Glucuronidase Type L-II (G8132): Effect of pH on Hydrolysis Efficiency

Incubation time: 2 hrs., Incubation temperature: 60 °C, Enzyme concentration: 10 kU/mL, Analyte concentration: 200 ng/mL



The experimental results summarized in **Table 3** present the optimal hydrolysis conditions for the specific compounds of each compound class. These results exhibited the highest level of hydrolysis for each specific compound, but may not be the optimal conditions for the entire class. In the case of benzodiazepine glucuronides, the *E. coli* enzyme did give optimal conversion for all benzodiazepines tested. In the case of the opioids, the hydrolysis conditions varied depending on the opioid glucuronide. In this case, the combination of β -glucuronidase/compound was much more compound specific than class general. The *E. coli* β -glucuronidase gave the best degree of conversion for the steroid and cannabinoid test analytes.

		Enzyme		Digestion	(hydrolysis) Conditions		
Class	Compound	Source	Cat. No.	рН	Temperature	Time	Units/mL
Benzodiazepines	Oxazepam	E. coli	G0799	6	60 °C	2 hrs	10,000
	Lorazepam	E. coli	G0799	6	60 °C	2 hrs	10,000
	Temazepam	E. coli	G0799	5	60 °C	2 hrs	10,000
Opioids	Norbuprenorphine	Bovine liver	G0376	5	60 °C	2 hrs	10,000
	Morphine	E. coli	G0799	6	60 °C	2 hrs	10,000
	Buprenorphine	P. vulgata	G8132	5	60 °C	2 hrs	10,000
	Hydromorphone	Bovine liver	G0251	5	60 °C	2 hrs	10,000
	Hydrocodone	Bovine liver	G0251	5	60 °C	2 hrs	10,000
Steroids	Testosterone	E. coli	G0799	6	60 °C	2 hrs	10,000
Pain Management	Tapentadol	E. coli	G0799	6	60 °C	2 hrs	10,000
Cannabinoids	(+)-11-nor-9-carboxy-Ƽ-THC	E. coli	G0799	6	60 °C	2 hrs	10,000

Conclusions

To optimize hydrolysis using β -glucuronidase, factors such as incubation time, temperature, hydrolysis pH, enzyme source, and enzyme concentration must be evaluated for each glucuronide metabolite to be analyzed. In the study reported here, the hydrolysis

pH was determined to be the most important factor for effective hydrolysis of glucuronide metabolites in urine. In some cases, a specific β -glucuronidase enzyme was effective for hydrolysis across a class of glucuronide metabolites. However, in most cases optimal hydrolysis conditions were compound dependent.

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For a general purpose enzyme, the aqueous solution of β -glucuronidase from limpets (*Patella vulgata*), Sigma catalog number G2174, gave the best overall conversion for all classes of compounds included in this study across all the pH conditions. This was not necessarily the optimal hydrolysis for each class or specific compound, but the limpet β -glucuronidase produced sufficient hydrolysis for all classes regardless of hydrolysis pH.

For optimal hydrolysis at specific β -glucuronidase and hydrolysis pH, the best conditions for overall conversion of all classes of compounds within this study were observed with the pH 5 conditions of the β -glucuronidase from bovine liver Type B-1, Sigma catalog number G0251.

Reference

1. McCarter, J. D.; Withers, S. G. Mechanisms of enzymatic glycoside hydrolysis. *Curr. Opin. Struct. Biol.* **1994**, *4*(6), 885-892.

Featured Products

Description	Cat. No.
Drug and Glucuronide Metabolite Test Compounds	
Steroids	
Testosterone glucuronide	NMID507
Testosterone sulfate	NMID508
Opioids	
Buprenorphine glucuronide	B-035
Buprenorphine	B-044
Norbuprenorphine glucuronide	N-045
Norbuprenorphine	N-059
Codeine glucuronide	C-087
Codeine	C-006
Codeine-6-β-D-glucuronide	C-126
Hydromorphone glucuronide	H-051
Hydromorphone	H-004
Oxymorphone glucuronide	O-030
Oxymorphone	O-004
Morphine glucuronide	M-031
Morphine	M-005
Benzodiazepines	
Lorazepam glucuronide	L-021
Lorazepam	L-901
Oxazepam glucuronide	O-023
Oxazepam	O-902
Temazepam glucuronide lithium salt	T-050
Temazepam	T-907
Pain Management	
Tapentadol-β-D-glucuronide	T-060
Tapentadol	T-058
Cannabinoids	
(+)-11-nor- Δ^9 -THC-9-Carboxylic acid glucuronide	T-038
(-)-11-nor-9-Carboxy-∆ ⁹ -THC	T-019

Description	Cat. No.
β-Glucuronidase Enzymes	
Bovine liver, Type B-1, solid	G0251
Bovine liver, Type B-3, solid	G0376
Burgundy snail (Helix pomatia) Type H-1, lyophilized powder	G0751
Burgundy snail (Helix pomatia) Type H-5, lyophilized powder	G1512
Escherichia coli, recombinant, aqueous glycerol solution	G0799
Escherichia coli, recombinant, lyophilized powder	G8295
Escherichia coli, Type IX-A, lyophilized powder	G7396
Escherichia coli, Type VII-A, lyophilized powder	G7646
Garden snail (Helix aspersa), Type HA-4, partially purified powder	G4259
Limpet (Patella vulgata), aqueous solution	G2174
Limpet (Patella vulgata), Type L-II, lyophilized powder	G8132
Solid Phase Extraction	
Supel [™] -Select SCX, 60 mg/3 mL, pack of 50 tubes	54241-U
Visiprep [™] SPE Vacuum Manifold, DL (Disposable Liner), 24-port model	57265
Visiprep [™] SPE Vacuum Manifold, DL (Disposable Liner), 12-port model	57044
Visiprep™ SPE Vacuum Manifold, standard, 24-port model	57250-U
UHPLC Columns	
Titan [™] C18, 5 cm × 2.1 mm l.D., 1.9 µm particles	577122-U
Solvents and Reagents	
Acetonitrile, LC-MS Ultra CHROMASOLV®, tested for UHPLC-MS, 1 L, 2 L	14261
Water, LC-MS Ultra CHROMASOLV®, tested for UHPLC-MS, 1 L, 2 L	14263
Ammonium formate, LC-MS Ultra eluent additive, 25 g	14266
Methanol, LC-MS Ultra CHROMASOLV®, tested for UHPLC-MS, 1 L, 2 L	14262
Ammonium hydroxide solution (28% NH_3 in H_2O , \geq 99.99% trace metals basis)	338818

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